

Chemical Protection From Effects (Cont.)

SOV/2206

- a) Prophylactic effect of some methemoglobin-forming agents 107
- b) Prophylactic effect of some tissue respiration inhibitors (cyanides, azides, nitriles) 111
- c) Prophylactic effect of substances easily reacting with oxygen 115
- d) Prophylactic effect occurring from inhaling carbon monoxide 116

Bibliography

- 5. Vitamins, enzymes, and their prophylaxis of radiation sickness 116
- Substances with Vitamin P function 118
- Pantothenic acid 123
- Acetylation coenzyme (KoA) 124
- Biotin and paraaminobenzoic acid 128

Bibliography

- 6. Various chemical compounds exhibiting protective effect against radiation of animals with X-rays and γ -rays 130
- Alcohols 131
- Carbohydrates (Sugars) 131
- Chelate compounds (complex-forming agents) 132
- Fatty acids 133
- Metals 133

Card 4/5

Chemical Protection From Effects (Cont.)

SOV/2206

Hormones
Narcotics

134
135

Bibliography

7. Some prospects of the development of chemical protection of
animals against the effect of ionizing radiation

135

136

Bibliography

141

AVAILABLE: Library of Congress

Card 5/5

TM/gmp
8/31-59

ROMANTSEV, Ye. F.; ZHUIANOVA, Z.I.

Acetylation of paraaminobenzoic acid in roentgen-irradiated rats.

Vop. med. khim. 5 no.1:10-15 Ja- F '59. (MIRA 12:3)

(ROENTGEN RAYS, effects,

on PARA acetylation in rats (Rus))

(PARAAMINO BENZOIC ACID, metab.

acetylation, eff. of x-rays in rats (Rus))

Chelation of Radioactive Isotopes from the Organism
(Chelation of Radioactive Isotopes from the Organism)
Moscow, 1970, pp. 1-151

The volume consists of a table of contents (attached), an introduction in which the author outlines the purpose of the book, and two sections. The first section deals with the problem of the chemical protection of the organism from penetrating radiation. A brief analysis is given of the contemporary state of the problem, data obtained in experiments are cited, and the theories of the mechanism of the protective action of some chemicals (antioxidants and pyridine derivatives) are examined. The second section deals with the problem of the elimination of radioactive isotopes from the organism. The effectiveness of certain chemicals which, when introduced into the organism, have the capacity to form with the isotopes stable compounds which would be readily eliminated from the organism is examined.

Introduction

5

Part 1. Chemical Protection from Ionizing Radiation

Present State of Chemical Protection from Ionizing Radiation, by V. S. Malobukha

7

Relationship Between the Structure and Properties of Sulfur-Containing Compounds and Their Protective Action from Penetrating Radiation, by V. G. Yakovlev

14

On the Mechanism of the Protective Action of Some Thiol Compounds, by V. G. Yakovlev and L. S. Isupova

41

Effect of Protective Doses of L-Cysteine on the Level of Superoxide Radicals in the Tissues of Rats Irradiated with X-Rays, by L. S. Isupova

55

Effect of Protective Substances on Protein Sulfhydryl Groups in the Organs and Tissues of Mice and Irradiated Animals, by V. G. Yakovlev and L. S. Isupova

62

Synthesis and Test of the Protective Action of a Series of Sulfur-Containing Compounds and Chemical Derivatives, by V. G. Yakovlev and V. S. Malobukha

72

Effect of Detachapropionylamide on the Formation of Organic Peroxides in the Irradiated Organism, by Ye. F. Kozlovskaya and G. I. Shilovskaya

82

Possibility of the Dilution of Chemical Compounds as Energy Traps in the Protection from Penetrating Radiation, by G. Ye. Fridkin

93

Part 2. Elimination of Radioactive Isotopes from the Organism

General Information

111

Physicochemical (Chemical) Investigation of the Effectiveness of Certain Complex-Forming Substances, by L. S. Malobukha and L. M. Pavlovskaya

112

Characteristics of the State of Radioactive Isotopes ^{32}P , ^{45}Ca and Co^{60} in the Blood, by L. M. Pavlovskaya and V. S. Malobukha

117

Effect of Complex-Forming Substances on the Binding Character of Radioisotopes in the Blood, by L. M. Pavlovskaya and V. S. Malobukha

125

Connective and Stability of γH Bond in Bone Tissue, by N. O. Pavlovskaya, O. L. Pavlovskaya, and V. S. Malobukha

130

Analysis of the Effectiveness of Complex-Forming Substances Which Hasten the Elimination of Radioactive Isotopes from the Organism, by G. Ye. Fridkin and V. F. Ushakov

135

REMANISE, 42

23

PHASE I BOOK EXPLOITATION SOV/5628

Akademiya nauk SSSR. Institut biologicheskoy fiziki

Rol' perekisey i kisloroda v nachal'nykh stadiyakh radiobiologicheskogo effekta (Role of Peroxides and Oxygen During Primary Stages of Radiobiological Effects) Moscow, 1960. 157 p. 4,500 copies printed.

Responsible Ed.: A. M. Kuzin, Professor; Ed. of Publishing House: K. S. Trinchin; Tech. Ed.: P. S. Kashina.

PURPOSE : This collection of articles is intended for scientists in radiobiology and biophysics.

COVERAGE: Reports in the collection deal with the role of peroxides and oxygen in the primary stages of a radiobiological effect. They were presented and discussed at a symposium held December 25-30, 1958, organized by the Institut biofiziki AN SSSR, (Institute of Biophysics, AS USSR). Twenty-eight Moscow scientists, radiobiologists, radiochemists, physicists, and

Card 1/5

23

Role of Peroxides and Oxygen (Cont.)

SOV/5628

physical chemists took an active part in the symposium. Between the time of its conclusion and the publication of the present book some of the materials were expanded. In addition to the authors the following scientists participated in the discussion: L. A. Tummerman, V. S. Tongur, G. M. Frank, Yu. A. Kriger, E. Ya. Grayevskiy, N. N. Demin, B. N. Tarusov, and I. V. Vereshchenskiy. References follow individual articles.

TABLE OF CONTENTS:

Kuzin, A. M. [Institut biologicheskoy fiziki AN SSSR - Institute of Biophysics, AS USSR]. Role of Formation of Peroxides During the Action of Radiation on Biological Specimens	3
Bakh, N. A. [Institut elektrokhimii AN SSSR - Institute of Electrochemistry, AS USSR]. Formation of Organic Peroxides Under the Action of Radiation	9
Dolin, P. I. [Institute of Electrochemistry, AS USSR]. Lifetime of Intermediate States Arising During the Action of Radiation on Aqueous Solutions	20

Card 2/5

Role of Peroxides and Oxygen (Cont.)	SOV/5628	7
Kolomiytseva, I. K., and A. M. Kuzin [Institute of Biophysics, AS USSR]. Lipid Peroxides in a Normal and in an Irradiated Animal Organism		26
Kuzin, A. M., L. M. Bronskaya, N. M. Berezina, and V. A. Yazykova [Institute of Biophysics, AS USSR]. Formation of Peroxides in Gamma-Irradiated Plant Seeds		33
Zhulanova, Z. I., I. A. Korovina, and Ye. F. Romantsev. Formation of Organic Peroxides in an Organism During Irradiation on an X-Ray Apparatus With a Dose Rate of 130 r/sec		43
Zhuravlev, A. I. Role of Antioxidants in Primary Radiobiological Effects		55
Mikhlin, D. M. (Deceased) [Institut biokhimii im. A. N. Bakha AN SSSR - Institute of Biochemistry imeni A. N. Bakh, AS USSR]. Effect of Ionizing Radiation of Oxidation-Reduction Reactions in a Cell		67
Card 3/5		

PLATE I FOR EXPLANATION

EXPLANATION

Radiation sickness; possible diagnosis; possible treatment (this book is intended for physicians and scientists). Moscow, 1950. 112 pages. 6,000 copies printed.

Eds.: A.I. Durnavay, Doctor and A.V. Iosadinsky, Professor; Chief Editor: N.A. Vlasova.

REMARKS: This textbook is intended for students in medical and biological sciences, and for physicians interested in the application of radiation sickness and medicine.

CONTENTS: This is a handbook on the application of radiation sickness and treatment of radiation sickness. It contains a large amount of material from foreign authors concerning radiation sickness and its treatment in Japan, and on clinical studies of radiation sickness in the U.S.A. Its presentation is logical, clear and concise.

Ch. III. Pathology of Radiation Sickness (Durnavay, A.I., Iosadinsky, A.V., Vlasova, N.A.).
Types of radiation sickness and special features of its development.
Radiation patterns in the development of acute radiation sickness.
Cause of death from ionizing radiation. Autopsy studies of radiation sickness.

General problems of pathogenesis
The nervous system
The cortex. Higher nervous activity
Vegetative centers. Spinal cord
Peripheral nervous system

Ch. IV. Irritation and Injury in Irradiated Animals (Durnavay, A.I., Iosadinsky, A.V., Vlasova, N.A.).
Infection problems in the exposed organs
Radiation sickness in the case of radiation sickness
Biological immunization reaction of exposed organisms

Effect of radiation on natural immunity
Artificial immunity of exposed organisms
Allergy in exposed organisms

Ch. V. Toxicology of Radioactive Substances (Iosadinsky, A.V., Professor)

Significance of physicochemical properties of radioactive substances
The ways radioactive substances enter the organism
Distribution of radioactive substances in the organism
Elimination of radioactive substances from the organism
Conditions influencing the nature of the effect of radioactive substances
Therapy of affection caused by radioactive substances

Ch. VI. Delayed Aftereffects of Affection Caused by Ionization

Ch. VII. Clinic for and Treatment of Radiation Sickness (Durnavay, A.I., Iosadinsky, A.V., Vlasova, N.A.).
Corresponding Member, Academy of Medicine USSR, and I.D. Glazunov, Professor

Acute radiation sickness
Therapy during radiation sickness
Chronic radiation sickness
Diagnosis of chronic radiation sickness

Ch. VIII. Utilization of Chemical Compounds to Protect Organisms from Ionization Radiation (Durnavay, A.I., Iosadinsky, A.V., Vlasova, N.A., Candidate of Biology)

Ch. IX. Pathologic Anatomy of Radiation Affection (Iosadinsky, A.V., Professor, Corresponding Member, Academy of Medicine USSR)

AVAILABLE: Library of Congress

Card 8/8

24/04/1950
9-4-50

ROMANTSEV, Ye.F.

Effect of L-cysteine and of β -mercaptoethylamine on oxygen content in venous blood. Med.rad. 5 no.2:19-20 F '60.

(MIRA 13:12)

(ETHYLAMINE) (CYSTEINE) (BLOOD-OXYGEN CONTENT)

ROMANTSEV, Ye.F.

Effect of certain protective substances on the oxygen requirements
of muscle tissue in rats. Med. rad. 5 no.4:86 Ap '60.

(MIRA 13:12)

(ETHYLAMINE)

(CYSTEINE)

(MUSCLE)

R. VANTSEV, YE. N. (USSR)

"Excretion of Dische-Positive Compounds after Irradiation
with Rapid Neutrons."

Report presented at the 5th International Biochemistry Congress,
Moscow, 10-16 August 1961

ROMANTSEV, Ye.F.; ZHULANOVA, Z.I.

Effect of protective agents and hypoxia on the formation of organic
peroxides. Radiobiologiya 1 no.1:73-77 '61. (MIRA 14:7)
(RADIATION PROTECTION) (ANOXEMIA)
(PEROXIDES)

30350

27.1220

S/205/61/001/004/008/032
D298/D303

AUTHOR: Romantsev, Ye. F.

TITLE: The excretion of Dishe compounds from the body with irradiation from fast neutrons and gamma-rays

PERIODICAL: Radiobiologiya, v. 1, no. 4, 1961, 508-511

TEXT: Since change in the DRN metabolism is one of the early biochemical responses to irradiation, the author studied the change in the excretion of Dishe-positive compounds with the urine in rats irradiated with fast neutrons in sublethal and lethal doses. An attempt was also made to determine the relative biological effectiveness of gamma-irradiation. Gamma-irradiation was effected from a Co^{60} source at an intensity of 5.1 r/min. in doses of 84.6, 253.9 and 507.9 r. Irradiation with fast neutrons was effected with the WPT-1000 (IRT-1000) reactor at the Institut atomnoy energii AN SSSR (Institute of Atomic Energy, AS USSR). The doses in this case were 84.6, 253.9 and

Card 1/2

30359

S/205/61/001/004/008/032

D298/D303

The excretion of...

507.9 r, at an intensity of 5.1 r/min. In the first few days after irradiation with gamma-rays, the excretion of Dishe-positive compounds with the urine increased in proportion to the total dose of radiation and reached its maximum (85% above normal) with a total dose of 507.9 r. At a dose of 84.6 r, no changes in the Dishe-compounds' content was noticed. At a dose of 253.9 r, Dishe-compound excretion rose by 52%. With fast neutron irradiation, the increase in the excretion was proportional to the total radiation dose, but ensued at much lower doses of absorbed energy: at 84.6 r it increased by 33% and at 253.9 r - by 96%. Further increase in the total radiation dose to 507.9 r caused no increase in the excretion of Dishe-positive compounds with the urine. There are 1 figure, 2 tables and 4 references: 3 Soviet-bloc and 1 non-Soviet-bloc. The reference to the English-language publication reads as follows: J. Pařizek, M. Arient, Z. Dienstbier, J. Skoda, Nature, 182, 1958.

SUBMITTED: April 7, 1961

Card 2/2

ROMANTSEV, Ye.F.; ZHULANOVA, Z.I.

Excretion of Dichet-positive compounds from the body in the utilization of substances for chemical protection from the action of ionizing radiations. Med. rad. 6 no.1:49-52 '61. (MIRA 14:3)
(RADIATION PROTECTION)

PETROV, R.V.; KOROGODIN, V.I.; LYASS, F.M.; NEYFAKH, A.A.; ROMANTSEV,
Ye.F.; VEREVKINA, N.M., red.; MORGUNOVA, G.M., tekhn. red.

[Contribution of radiology to the development of the medical
and biological disciplines]Vklad radiologii v razvitie mediko-
biologicheskikh distsiplin. [By] R.V.Petrov i dr. Minsk, Izd-
vo M-va vysshego, srednego spetsial'nogo i professional'nogo
obrazovaniia BSSR, 1962. 145 p. (MIRA 15:9)
(RADIOBIOLOGY) (RADIOLOGY, MEDICAL)

PHASE I BOOK EXPLOITATION

SOV/6407

Romantsev, Yevgeniy Fedorovich

Radiatsiya i khimicheskaya zashchita (Radiation and Chemical Protection) Moscow, Gosatomizdat, 1963. 206 p. 6500 copies printed.

Ed.: A. V. Matveyeva; Tech. Ed.: Ye. I. Mazel'.

PURPOSE: This book is intended for specialists in a variety of fields who are interested in present-day problems in radiobiology and for doctors and students.

COVERAGE: The book studies the mechanisms by which chemical radioprotective agents counteract the effects of ionizing radiation and analyzes the primary biochemical and physicochemical processes which accompany the irradiation of living cells. Of particular interest are specific reaction links and mechanisms which are common to more than one chemical radioprotector.

Card 1/7

Radiation and Chemical Protection

SOV/6407

Compounds such as vitamins and hormones, the mechanism of whose action increases the general radioresistance of the organism, occupy a special position among chemical radioprotectors. Separate chapters deal with the primary physicochemical processes which occur with irradiation, including disturbances of the correlation of biochemical processes in cellular microstructures, and examine the relation of these processes to the protective mechanisms. Radioprotective agents have important applications for medicine, the atomic energy industry, and spaceflight. References are listed following each chapter. There are a total of 527 citations, of which approximately 170 are from Russian, 290 from English, 30 from French, 20 from German, 9 from Italian, and 8 from other sources. No personalities are mentioned.

Card 2/7

ROMANTSEV, Ye.F.

New data on the mechanism of protector action. Trudy MOIP. Otd.
biol. 7:102-107 '63. (MIRA 16:11)

ROMANTSEV, Ya.F.; TIKHOMIROVA, M.V.

Protection of animals from gamma rays with the help of some
aminophenones. Radiobiologiya 3 no.1:126-129 '63.

(MIRA 16:2)

(GAMMA RAYS--SAFETY MEASURES) (PROPIOPHENONE)
(ACETOPHENONE) (BUTYROPHENONE)

GROMOV, V.A.; ZHULAIKVA, Z.I.; ROMANTSEV, Ye.F.; SKOLIN, D.D.; SOKOLOVA, G.N.

Changes in the composition of liver lipid fractions in animals
exposed to radiation. Radiobiologiya 4 no.3:378-380 '64.

(MIRA 17:11)

1. KULIKOV, Y. P., and KURBANOV, N. I.

Mechanism of the radioprotective activity of ortho-, meta- and paraaminophenolphenones. Radiobiologiya 6: 510-515 (1971).

(USSR: 1972)

2. Institut biokhimi Ministerstva zdorovokhrazheniya SSSR, Moscow.

KALISTRATOV, G.V.; ROMANTSEV, Ye.F. (Moskva)

Distribution and elimination of S^{35} -labeled β -mercaptopyrrolamine
from the organism of mice. Farm. i toks. 27 no.3:364-367 My-Je '64.
(MIRA 18:4)

KALISTRATOV, G.V.; ROMANTSEV, Ye.F. (Moskva)

Nature of the bonds between β -mercaptoethylamine and the
proteins of liver cell nuclei. Vop. med. khim. 11 no.4:
38-41 J1-Ag '65. (MIRA 18:8)

ACC NR: AM6026752

Monograph

UR/

Romantsev, Yevgeniy Fedorovich; Blokhina, Vera Dmitriyevna;
Koshcheyenko, Nikolay Nikolayevich; Filippovich, Igor' Vladimorovich

Early radiation and biochemical reactions (Ranniye radiatsionno-biokhimicheskiye reaktsii) Moscow, Atomizdat, 1966. 270 p. illus., biblio., tables. 2200 copies printed.

TOPIC TAGS: radiation biochemistry^{cal effect}, radiobiology, ~~radiation biology~~
~~effect~~, radiation chemistry, ~~radiation sensitivity~~, ~~radiation~~
~~resistance~~, radiation sickness, ~~chemical radiation protection~~,
DNA, RNA, ~~anti radiation drug~~, ~~radiation cell effect~~

PURPOSE AND COVERAGE: This book is intended for biologists and biochemists concerned with problems of radiation biochemistry. The authors investigate the nature of early biochemical changes in the living cell following irradiation, and the effect of protective chemicals used in counteracting radiation in the living organism. The formation of peroxides and peroxide-like compounds, the effect of radiation on the synthesis of DNA and information RNA, the formation of macroergs, and other radiation problems in radiation biochemistry are discussed. An attempt is also made to determine the relationship between the operational mechanism of several protective chemicals and "radiosensitive" biochemical reactions. Each chapter is accompanied by an extensive list of references.

Card 1/2

UDC: 613.541.15:543.9.

ACC NR: AM6026752

TABLE OF CONTENTS [abridged]:

Editor's foreword -- 3

Ch. 1. Formation of H_2O_2 , organic peroxides, and peroxide-like compounds due to irradiation -- 7

Ch. 2. Effect of radiation and chemical protectives on DNA biosynthesis -- 43

Ch. 3. Effect of ionizing radiation on the biosynthesis of RNA and the possibility of normalizing it with protectors -- 120

Ch. 4. Effect of ionizing radiation on the synthesis of albumen -- 195

Ch. 5. Oxidative phosphorylation in the tissue of radiated animals -- 244

SUB CODE: 06/ SUBM DATE: 24Jan66/ ORIG REF: 336/ OTH REF: 805

Card 2/2

S/056/62/043/001/047/056
B102/B104

AUTHORS: Barkov, L. M., Mukhin, K. N., Ogurtsov, V. V.,
Romantseva, A. S., Svetlilobov, I. A., Chuyeva, S. A.,
Shlyapnikov, R. S., Likhachev, M. F., Stavinskiy, V. S.,
Strunov, L. N.

TITLE: The problem of the D^+ -meson

PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, v. 43,
no. 1(7), 1962, 335-337

TEXT: The authors have searched for a D^+ -meson production or a decay among 14,000 pairs of photographs. A propane bubble chamber with pulsed magnetic field was irradiated with a beam of positively charged particles (momentum ≈ 1.8 Bev/c) containing up to 9% K^+ mesons. The processes looked for were $K^+ + p \rightarrow D^+ + \Sigma^+$ and

$D^+ \rightarrow K^0 + \pi^+$

$D^+ \rightarrow K^+ + \pi^0$

Card 1/2

The problem of the D^+ -meson

S/056/62/043/001/047/056
B102/B104

The first branch of the decay reaction is the more possible. Neither a process $K^+ + p \rightarrow D^+ + \Sigma^+$ nor one of the type $K^+ + n \rightarrow D^+ + \Sigma^0$ could be found. It is inferred that the D^+ meson production cross section in K^+N reactions will be smaller than $1.2 \cdot 10^{-29} \text{ cm}^2$.

ASSOCIATION: Institut atomnoy energii (Institute of Atomic Energy)
(R. S. Shlyapnikov); Ob"yedinennyy institut yadernykh
issledovaniy (Joint Institute of Nuclear Research)
(L. N. Strunov)

SUBMITTED: April 25, 1962

Card 2/2

ROMANTSEVA, A.S.

315h1
S/627/60/002/000/024/027
D299/D304

3.24/0 (1205, 2705, 2805)

AUTHORS: Varfolomeyev, A. A., Gerasimova, R. I., Gurevich, I.I.,
Makar'ina, L.A., Romantseva, A. S., and Chuyeva, S. A.

TITLE: Electron-photon showers with energies of 11^{11} - 10^{13} ev.
in nuclear emulsions

SOURCE: International Conference on Cosmic Radiation. Moscow,
1959. Trudy. v. 2. Shirokiye atmosferynye livni i kas-
kadnyye protsessy, 299-306

TEXT: A detailed investigation was carried out of 15 electron-photon showers with energies $>10^{11}$ ev., at low depths. In contradistinction to other works, the results are compared with those obtained for cascades by the Monte Carlo method. Six emulsion stacks were used, with total volume of about 10 liters. In 5 of the stacks of emulsion Р-НИКФН (R-NIKFI), the grain density of relativistic electrons was 30 - 35 grains per 100 μ . The energy E_γ of primary quanta which generate the shower, was determined from the

Card 1/4

Electron-photon showers ...

315h1
S/627/60/002/000/024/027
D299/D304

number of cascade electrons of energy higher than $\mathcal{E}_c = 300$ Mev, at a depth of $2.5 - 3.0 t_0$. A table lists (for comparison) the values of E_f , obtained by the Monte Carlo method and by formula

$$R = \frac{1}{16,1} \left\{ 45,0 + \ln \left[\left(\frac{2x}{E} \right)^2 (1 + 140 x) \right] \right\} \quad (1)$$

where x is the distance from the pair vertex in cm; this formula is semiempirical and represents the ratio of ionization losses of pairs to those of relativistic electrons; the ionization losses are due to mutual shielding of electron and positron fields. In the experiments, particular care was taken to detect the vertices of the electron-positron pairs, formed at depths $\leq 1.5 t_0$. After determining the lateral shower distribution, the energy of the electrons of the pairs was measured by means of multiple scattering (to an accu-

Card 2/4

Electron-photon showers ...

31541
S/627/60/002/000/024/027
D299/D304

racy of 20 - 30%) for energies of up to $(5-7) \cdot 10^8$ ev. The total number of pairs formed at depths $\leq 1.0 t_0$ and $\leq 1.5 t_0$ with energies higher than (1-2) Mev, is plotted in two figures, from which it is evident that the experimental points fit better the curve which takes into consideration the influence of the medium on the bremsstrahlung (the curve obtained by Migdal's formula); the curve obtained by Bethe-Heitler's formula does not fit the experimental results. The figures also show that not one of the 15 showers under consideration is anomalous. Apparently, the majority of so-called "anomalous" showers, described in literature, can be explained by statistical fluctuations in the cascades or by improper determination of the energy of primary electron-positron pairs. Another figure exhibits the experimental curves of longitudinal shower development; here, too, no appreciable deviations from the corresponding theoretical curves are observed. A table lists data on the number of pairs formed at small distances $r < 0.5 \mu$ from the nearest electron track; these data might be useful in analyzing the cross-section for pair formation by high-energy electrons. There are 4

Card 3/4

Electron-photon showers ...

315h1
S/627/60/002/000/024/027
D299/D304

figures, 3 tables and 21 references: 10 Soviet-bloc and 11 non-Soviet-bloc. The 4 most recent references to the English-language publications read as follows: K. Pinkau. Nuovo Cim., 3, 1285, 1956; H. Fay. Nuovo Cim., 5, 293, 1957; J. Iwadare. Phil. Mag., 3, 680, 1958; S. K. Srivivasan, J. S. Butcher, B. A. Chartres, H. Messel. Nuovo Cim., 9, 77, 1958.

Card 4/4

L 58337-65 EWI(m) Feb DIAAP
ACCESSION NR: AT5010445

UR/3136/64/000/700/0001/0015

AUTHOR: Kruchinin, S. P.; Mukhin, K. N.; Romantseva, A. S.; Svetloloobov, I. A.;
Sulkovskaya, M. M.; Chuyeva, S. A.; Shlyapnikov, R. B. 28
BH

TITLE: Elastic p-p scattering at 1.45 BeV

SOURCE: ⁷⁹Moscow. Institut atomnoy energii. Doklady, no. 700, 1964. Uprugoye
(p-p)-rasseyaniye pri 1,45 BeV, 1-15

TOPIC TAGS: elastic scattering, proton proton scattering, pion scattering, differential cross section

ABSTRACT: A propane bubble chamber was used to investigate the angular dependence of elastic scattering of protons by protons at an incident-proton momentum of 2.2 BeV/c, which is higher than the energies used in earlier investigations. The protons came from the 10 BeV accelerator of the Joint Institute of Nuclear Research. A total of 17,000 pairs of stereophotographs was scanned, ~ 900 cases of elastic π^+p and pp scattering cases were analyzed, and the reduction of these data made it possible to determine the differential cross section of elastic pp scattering at 1.45 BeV over the entire angle interval of 0-90° (c.m.s.). Calculations based on

Card 1/2

L 58337-65

ACCESSION NR: AT5010445

the optical model with a small region of phase shift gave best agreement with the experimental data with parameter values $R_1 = 0.45 F$, $R_2 = 0.95 F$, $a = 0.344$, and $\varphi = 1.77$ rad. (R --interaction radius, φ --phase shift, a --amplitude of transmitted wave for a unit amplitude of incident wave). In the energy region from 0.38 to 30.9 BeV, the differential cross section is proportional to $A \exp(-P_{c.m.s.}/P_0)$, with $A = 115$ mb/sr and $P_0 = 143$ MeV/c. "The authors thank I. I. Gurevich for valuable advice, A. P. Benediktov, V. I. Baranov, and A. V. Tel'nov for help in operating the equipment, and V. S. Balova, L. S. Baturina, and A. A. Kondrashina for participating in the measurements." Orig. art. has: 5 figures, 9 formulas, and 1 table.

ASSOCIATION: none

SUBMITTED: 00

ENCL: 00

SUB CODE: NP

NR REF SOV: 001

OTHER: 012

BR
Card 2/2

VARFOLOMEYEV, A.A.; GERASIMOVA, R.I.; MAKAR'INA, L.A.; ROMANTSEVA, A.S.;
CHUYEVA, S.A.

Ionization along the track of a high-energy electron-positron
pair. Zhur.eksp. i teor.fiz. 36 no.3:707-716 Mr '59.

(MIRA 12:5)

(Ionization) (Photography, Particle track)

ROMANTSEVA, A.S.

"ELECTRON-PHOTON CASCADES WITH ENERGIES FROM 10^{11} TO 10^{13} ev IN NUCLEAR EMULSIONS"

A.S. Romantseva, A.A. Varfolomeyev, R.I. Gerasimova, I.I. Gurevich, L.A. Makaryina
S.A. Chuyeva

Fifteen electron-photon cascades with energies from 10^{11} to 10^{13} ev, recorded in six emulsion stacks with a total volume of 10 l, have been investigated.

The energies of the primary photons evoking the cascades were determined by the energy spectrum of the cascade electrons at a depth of $2.5 \pm 3t_0$ (t_0 - rad. unit).

The grain density and the gap density were measured for the first pairs. In all the pairs with energies 3×10^{11} ev, a decrease in grain density at the apex caused by the screening effect was discovered.

The following experimental relation of the ionization losses of pair (1) was obtained:

where I_{per} is the specific ionization electron loss at the ionization plateau, x is the distance from the apex of the pair in cm, and E is the energy of the photon which produced the pair.

The expression obtained for $I/2I_{pe}$ may be used to determine the E energy from experimental values for I . An estimation of the E error is given, taking into consideration the screening effect.

The number of electron-positron pairs produced at depths of $1.0t_0$ and $1.5t_0$ was measured.

ROMANTSEVA, A.S. (CONTINUED)

The results agree with the calculated data obtained by the Monte Carlo method, taking into consideration the effect of the medium of Bremsstrahlung (laundau-Pomeranckuk and Ter-Mikaelyan effects).

report presented at the International Cosmic Ray Conference, Moscow 6-11 July 1959

For 10 cascades with $E.1.8 \times 10^{11}$ ev, the probability of $p ()^2$ from the criterion / 2, is 2.5 * when compared with the curves which do not consider the effect of the medium, and 80-95% when compared with the calculations that take into consideration the effect of the medium on the Bremsstrahlung.

VARFOLOMEYEV, A.A.; GERASIMOVA, R.I.; GUREVICH, I.I.; MAKAR'INA, L.A.;
ROMANTSEVA, A.S.; CHUYEVA, S.A.

Effect of the density of the medium on bremsstrahlung in electron-
photon showers involving energies from 10^{11} to 10^{13} ev. Zhur.
eksp. i teor. fiz. 38 no.1:33-45 Jan '60. (MIRA 14:9)
(Bremsstrahlung) (Cosmic rays)

6
BARKOV, L.M.; MUKHIN, K.N.; OGURTSOV, V.V.; ROMANTSEVA, A.S.; SVETLOLOBOV,
I.A.; CHUYEVA, S.A.; SHLYAPNIKOV, R.S.; LIKHACHEV, M.F.; STAVINSKIY,
V.S.; STRUNOV, L.N.

The problem of the D^+ -meson. Zhur. eksp. i teor. fiz. 43 no.1:335-
337 J1 '62. (MIRA 15:9)

1. Ob"yedinennyy institut yadernykh issledovaniy (for Strunov).
(Mesons)

Romantseva, A. S.

Angular distribution of fragment from the fission of uranium-235 by neutrons of different energies. A. A. Vartolomeev, A. S. Romantseva, and V. M. Kutukova, *Doklady Akad. Nauk S.S.S.R.* 105, 093-5 (1955).—The angular distribution was detd. of fission fragments from the natural mixt. of U isotopes interacting with neutrons of ~4, 12, 14, and 17-m.e.v. energy. A correlation is observed between the direction of scattering of fission fragments from heavy nuclei and the direction of flight of the particles causing fission. The liquid-drop model cannot account for this. On the Hill and Wheeler collective model of the nucleus, transmission of the energy of the impinging particle to surface vibrations of the nucleus (across "wall" of nucleus) is possible, and in principle there can exist a direct connection between the momentum of the impinging particle and the direction of scattering of the fragments. A detn. of the dependence of the anisotropy of the fission fragment on the energy of the particle causing fission and on the fissioning nucleus is of significance for the mechanism of fission.

J. J. Mitchell

ROMAN 75-10, H. 5

21(6)
 AUTHORS: Vasyukov, A. A., Gerasimova, R. I., Kacarina, L. A.,
 Romantsev, A. S., Chuyeva, S. A.
 TITLE: Ionization Along the Tracks of Electron-Positron Pairs of
 High Energy (Ionizatsiya vol. slabo elektronno-positronnykh
 par vysokoy energii)
 PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, 1959,
 Vol. 36, No. 3, pp 707-716 (US28)

ABSTRACT: In the introduction the authors discuss the problem and the
 results of several already published works dealing with
 this subject. Table 1 contains for the 5 investigated energies
 (R-5), 0-202, D-84, D-44 and I-102) the data of the emulsion
 plates in which they were recorded (see previous paper by the
 same authors, reference 7); table 2 contains a list of the
 K_{α} -values according to Janossy (Janossi) (Refs 10, 12) and
 according to Chudakov (Ref 1). (Today it is possible to ob-
 tain more exact K_{α} -values from curves by the Monte Carlo
 method by taking into account the influence exercised by matter on brems-
 strahlung into account. The publication of respective results
 has been announced). A very detailed chapter of this paper
 deals with gauging of the emulsions (type R-XIP1). The follow-

Card 1/3

Ionization Along the Tracks of Electron-Positron Pairs of High Energy
 ing experimental data concern the track densities of five
 high-energy electron-positron pairs in these emulsions. Measure-
 ments were carried out on the first pairs of electron-photon
 showers. Pair energy was determined from the energy spectrum
 of the cascade electrons at a distance of 3-5 - 3 radiation
 lengths from the vertex of the first pair. In three cases
 pair energy was nearly 10^{12} ev and in two cases it was
 approximately $3 \cdot 10^{11}$ ev. Track density was determined by two
 methods: from the grain density in the track and from the gap
 length distribution coefficient. Compared with a particle
 for which the specific energy loss is twice as great as the
 ionization loss of the electron, the track density of the
 pair near the vertex was found to be smaller. This decrease
 of the pair track density can be explained by the mutual
 screening of the electron and positron during ionization. The
 results obtained are compared with the theoretical ionization
 curves for pairs calculated by A. Fe. Chudakov (Ref 1). The
 authors finally thank Professor I. I. Gurevich for his inter-
 est and discussions, A. A. Kondrashina for his help in

Card 2/3

Ionization Along the Tracks of Electron-Positron Pairs of High Energy
 evaluating measuring results, and D. V. Sazonov and his
 group for developing the plates of emulsion plates. There
 are 8 figures, 2 tables, and 21 references, 3 of which are
 Soviet.

SUBMITTED: August 18, 1958

Card 3/3

SHKLYAR, F. R.; TIMOFEYEV, V. N.; Prinimali uchastiye: PAKHALUYEV,
K. M., inzh.; KOROLEV, N. M., inzh.; CHEREMNYKH, V. I.,
laborant; GERASIMOV, G. I., laborant; ROMANTSEVA, E. P.,
laborant; RUZHENTSEVA, T. M., laborant

Experimental investigation of the regenerative heat exchange
process. Sbor. nauch. trud. VNIIMT no.8:119-136 '62.
(MIRA 16:1)

(Air preheaters--Testing)
(Heat--Transmission)

STEPANOV, Georgiy Yur'yevich. Prinimali uchastiye: SIROTKIN, Ya.A.;
NAUMOVA, L.G.; ROMANTSEVA, L.I.; SHUSTOV, S.N., red.;
BRUDNO, K.F., tekhn. red.

[Fluid dynamics of turbomachine cascades]Gidrodinamika reshe-
tok turbomashin. Moskva, Fizmatgiz, 1962. 512 p.

(MIRA 15:8)

(Turbomachines--Fluid dynamics)

(Cascades(Fluid dynamics))

1235

[illegible]

0-5162-3009-0/00/0000-0000

The authors have shown that the effect of the shift on the frequency of the shift is not only a function of the shift itself, but also of the frequency of the shift. The authors have shown that the effect of the shift on the frequency of the shift is not only a function of the shift itself, but also of the frequency of the shift. The authors have shown that the effect of the shift on the frequency of the shift is not only a function of the shift itself, but also of the frequency of the shift.

[illegible]

CONCLUSIONS

ROMANTSEVA, L.M.; SOLODKIN, I.S.

Corrosion of nitrided steel by the combustion products
of sulfur fuels. Uch. zap. MGPI 99:87-93 '57.

(MIRA 12:3)

(Steel--Corrosion) (Sulfur compounds)

SOV/137-58-12-24821

Translation from: Referativnyy zhurnal. Metallurgiya, 1958, Nr 12, p 118 (USSR)

AUTHORS: Romantseva, L. M., Solodkin, I. S.

TITLE: Corrosion of Nitrogenized Steel by Products of Combustion of Sulfur-containing Fuel (Korroziya azotirovannoy stali produktami sgoraniya sernistogo topliva)

PERIODICAL: Uch. zap. Mosk. gos. ped. in-ta, 1957, Nr 99, pp 87-93

ABSTRACT: An investigation was made of the effect of the S content of fuel (F) on the corrosive activity of its combustion products (CP) on nitrogenized 38KhMYuA steel. Experiments were performed with F containing 0.11, 1.15, and 2.2% S. Specimens were subjected to the action of CP for 10 min, whereupon they were placed into a moisture chamber and left there first for one day and then for four days. It was found that the corrosive aggressiveness of CP increased with the increase of the S content in the F. An investigation was made of the dissolution of the nitrogenized layer of 38KhMYuA steel in H_2SO_4 , H_2SO_3 , H_2CO_3 , HNO_3 , formic, acetic, and butyric acids and in their mixtures which can be formed in CP. The experiments were carried out for 10 hours and the concentration of the acids was 0.5N. It was established that

Card 1/2

SOV/137-58-12-24821

Corrosion of Nitrogenized Steel by Products of Combustion of Sulfur (cont.)

the most corrosive component of the CP is H_2SO_4 . Dethylamine, diethylaniline, butylphosphate, and hexylphosphate (1% by volume) were added to the sulfur-containing F, and the effect of these additions on the corrosion of nitrogenized steel in the CP was studied. It was established that additions of amines and phosphates to F slow down greatly the development of corrosion of 38KhMYuA steel in the CP of sulfur-containing F.

G. M.

Card 2/2

BEKIN, I.M.; REZANTSEVA, L.M.; SUKHANOVA, V.A.; COLOSHCHEPOV,
I.M., Eds.

[Plastics in armored equipment:] Plastmassy v oruzhetankovoi
tekhnike. Moskva, Voenizdat, 1965. 136 p. (EIRA 18:9)

ROMANTSEVA, L.M.; KOZLOVA, I.V.

Chemical processes on electrodes during electrolysis. Khim. v shkole
13 no.5:25-40 S-O '58. (MIRA 11:9)
(Electrochemistry)

L 07102-07 EWP(j)/EWT(m) RM
ACC NR: AP6028198

SOURCE CODE: UR/0189/66/000/002/0080/0082

AUTHOR: Romantseva, T. I.; Gromova, M. I.; Peshkova, V. M.

ORG: Analytical Chemistry Department (Kafredra analiticheskoy khimii)

TITLE: Study of complexing in the system neodymium - 5,7-dichlorooxine - amyl acetate - water

SOURCE: Moscow. Universitet. Vestnik. Seriya II. Khimiya, no. 2, 1966, 80-82

TOPIC TAGS: neodymium compound, stability constant, chelate compound

ABSTRACT: Complexing between Nd³⁺ and 5,7-dichlorooxine (Cl₂OX) was studied in the system Nd - Cl₂OX - amyl acetate - H₂O. The overall and successive stability constants of the complex formed were calculated by the methods of (1) Bjerrum (graphically and by computation) and (2) Dyrssen and Sillen. The complex was found to have the composition Nd(Cl₂OX)₃. The overall stability constant is -4.37, whereas the corresponding constant for the Nd - Cl₂OX - chloroform - H₂O system is -4.58. This indicates that the advantage of amyl acetate over chloroform in extracting dichlorooxine complexes of rare earths lies in the fact that the extraction with amyl acetate begins in a more acidic medium, i. e., under conditions where the rare earth ion cannot hydrolyze. Orig. art. has: 3 figures, 1 table and 1 formula.

SUB CODE: 07/ SUBM DATE: 21Jul65/ ORIG REF: 002/ OTH REF: 004

Card 1/1 hpc

UDC: 543.7 546

ACC NR: AP6013827

SOURCE CODE: UR/0189/65/000/006/0074/0078

AUTHOR: Romantseva, T. I.; Gromova, M. I.; Peshkova, V. M.

ORG: Chair of Analytical Chemistry, Moscow State University (Kafedra analiticheskoy khimii, Moskovskiy gosudarstvennyy universitet)

TITLE: Spectrophotometric determination of neodymium in the presence of praseodymium and samarium

SOURCE: Moscow. Universitet. Vestnik. Seriya II. Khimiya, no. 6, 1965, 74-78

TOPIC TAGS: spectrophotometric analysis, neodymium, praseodymium, samarium, absorption spectrum, organometallic compound

ABSTRACT: Absorption spectra of complex compounds formed by neodymium, samarium and praseodymium with 7-iodo-8-hydroxyquinoline-5-sulfonic acid were studied. A strong hypsochromic shift of the absorption band of this reagent in alkaline media, combined with the bathochromic shift of the characteristic absorption maximum of the complex formed by this reagent with neodymium, creates favorable conditions for determining neodymium at $\lambda_{\max}$ 581 m μ . Neodymium was determined in Nd-Pr and Nd-Sm pairs in water-alcohol solutions at pH 8.0-11.0, using two wavelengths: λ_1 581 m μ and λ_2 590 m μ . The data show that the method used permits a sufficiently accurate determination of neodymium in praseodymium and samarium in amounts of 0.0043-0.043 mg/ml (or $3 \cdot 10^{-5}$ -

UDC: 543.7

Card 1/2

L 30227-66

ACC NR: AP6013827

- $1 \cdot 10^{-4}$ M) at Nd:Pr(Sm) ratios of 1:99 to 1:8. Orig. art. has: 4 figures, 2 tables, 3 formulas.

SUB CODE: 07/

SUBM DATE: 22Mar65/

ORIG REF: 009/

OTH REF: 005

Card 2/2

110

L 34074-06 ENT(M)/EAT(1)/FT1 TJP(c) JD/JG

NR: AP6010716

SOURCE CODE: UR/0189/66/000/001/0073/0078

AUTHOR: Romantseva, T. I.; Gromova, M. I.; Peshkova, V. M.

ORG: Analytic Chemistry Department, Moscow State University (kafedra analiticheskoy khimii, Moskovskiy gosudarstvennyy universitet)

TITLE: Use of different variants of spectrophotometric measurements in the determination of erbium in holmium and thulium and of ytterbium in lutetium

SOURCE: Moscow. Universitet. Vestnik. Seriya II. Khimiya, no. 1, 1966, 73-78

TOPIC TAGS: spectrophotometric analysis, erbium, holmium, thulium, ytterbium, lutetium

ABSTRACT: Continuing a study of the spectrophotometric analysis of rare earth mixtures for their individual components, the authors have compared different variants of the spectrophotometric method and attempted to show their applicability to the analysis of separate pairs of rare earth elements, taking as an example complex compounds of a series of elements of the yttrium subgroup with 7-iodo-8-hydroxyquinoline-5-sulfonic acid. Absorption spectra of water-alcohol solutions of complexes of erbium, holmium, thulium, ytterbium, and lutetium were studied. Erbium was determined in Er-Ho and Er-Tm pairs, and a statistical treatment of the results showed that the errors had a random character in the case of the Er-Tm pair, whereas in the case of Er-Ho the results were systematically high. However, Er was reliably determined in Ho by also

Card 1/2

UDC: 543.7

L 34374-66

ACC NR: AP6010716

making use of a differential method. Ytterbium was determined in the Yb-Lu pair; mathematical treatment of the results showed the errors to be random in character, but the standard deviation was fairly appreciable. The methods employed permitted the determination of Er in amounts from 3 to 10% ($9 \times 10^{-5} \text{ M} - 3 \times 10^{-4} \text{ M}$) in Tm and from 5 to 10% ($1.5 \times 10^{-4} \text{ M} - 3.0 \times 10^{-4} \text{ M}$) in Ho, and the determination of Yb in amounts from 15 to 30% ($4.5 \times 10^{-4} \text{ M} - 9.0 \times 10^{-4} \text{ M}$) in Lu. Orig. art. has: 6 figures and 4 tables.

SUB CODE: 07/ SUPM DATE: 02Apr65/ ORIG REF: 006

Card 2/2

L 29946-65 EWT(m)/EWP(t)/EWP(b) IJP(o) JD/JG

ACCESSION NR: AP4044080

S/0189/64/000/004/0057/0061

AUTHORS: Gromova, M.I.; Romantseva, T. I.; Peshkova, V. M.

TITLE: Using the absorption spectra of the dichloroxinates of praseodymium, neodymium and samarium for the determination of these elements

SOURCE: Moscow. Universitet. Vestnik. Seriya. 2. Khimiya, no. 4, 1964, 57-61

TOPIC TAGS: praseodymium, neodymium, samarium, spectrophotometric determination, dichlorohydroxyquinoline, rare earth dichloroxinate, extraction, coefficient of extinction, absorption spectrum

ABSTRACT: The spectrophotometric determination of praseodymium, neodymium and samarium, or mixtures of these, complexed with 5,7-dichlorohydroxyquinoline was investigated. Optimum extraction of these complexes from aqueous solutions with chloroform is in the pH 6.5 -7 to 8.5-10 range; below pH 6.5 complex formation is in progress. The rare earth salts were dissolved in dilute HClO_4 , the reagent was

Card 1/3 71

L 29916-65

ACCESSION NR: AP4044080

used as a 0.1% solution in 3N HCl. The absorption spectra of the rare earth dichloroxinates were obtained. All three complexes absorb strongly in the 395-400 millimicron region; Nd has several peaks, while Pr and Sm show no peaks in the 500-850 millimicron region; the Nd maximum at 581 millimicrons is most pronounced; Sm has a maximum at 1085 millimicrons. Concentrations of the elements in Nd-Pr and Nd-Sm mixtures were determined by solution of the equation $D = \sum_{a=1}^n \epsilon \lambda_a c_a$, where $n = 2$, $\lambda_1 = 581$ and $\lambda_2 = 640$ millimicrons for Nd-Pr, and $\lambda_1 = 581$ and $\lambda_2 = 1085$ millimicrons for Nd-Sm mixtures. The coefficients of extinction ϵ for the selected wave lengths (in millimicrons) for the various complexes: Pr, at $\lambda = 581$, $\epsilon = 10.34$ and at $\lambda = 640$, $\epsilon = 7.85$; Sm, at $\lambda = 1085$, $\epsilon = 5.0$ and at $\lambda = 581$, $\epsilon = 3.6$; Nd, at $\lambda = 581$, $\epsilon = 63.0$ and at $\lambda = 640$, $\epsilon = 7.8$. Because of the limited solubility of these rare earth complexes in chloroform, the determinable concentration of Nd in Pr and Sm is in the 1×10^{-4} to 5×10^{-4} M range (0.014-0.072 mg/ml of Nd in the presence of 0.12-0.07 mg/ml of Pr or 0.13-0.075 mg/ml of Sm). Pr and Sm cannot be determined in the presence of relatively large amounts of Nd. Orig. art. has: 2 tables and 3 figures.

Card 2/3

L 29946-65

ACCESSION NR: AP4044080

ASSOCIATION: MGU, Kafedra analiticheskoy khimii (Moscow State
University, Department of Analytical Chemistry)

SUBMITTED: 24Mar64

ENCL: 00

SUB CODE: 1C,OP

NR REF SOV: 001

OTHER: 009

Card 3/3

ROYNISHVILI, N. N.

1-100

Distribution of charged particles in electron nuclear
showing according to their momenta. L. D. Gela-
vanishvili, Z. Sh. Mamalashvili, N. N. Roishvili, B. I.
Tsagareli, A. I. Tsintsadze, and G. B. Chikvashvili. *Bull.*
Acad. Sci. U.S.S.R., Phys. Ser. 6, 677-8(1965)(English
translation).—See *C.A.* 50, 7018g. *N. M. R.*

72ue

sci

BMZ

GEDEVANISHVILI, L.D.; MANDZHAVIDZE, Z.Sh.; ROYNISHVILI, N.N.; TSAGARELI, E.I.
TSINTSABADZE, A.I.; CHIKOVANI, G.Ye.

Pulse distribution of charged particles in electronic and nuclear
showers. Izv. AN SSSR. Ser. fiz. 19 no.6:748-749 N-D '55. (MLRA 9:4)

1. Institut fiziki AN Gruz.SSR i Tbilisskiy gosudarstvennyy universi-
tet imeni I.V.Stalina.

(Cosmic rays) (Nuclear physics)

ROYNISHVILI, N.N.

AUTHOR: MANDZHAVID Z E, Z. SH., ROYNISHVILI, N. N., CHIKOVANI, G. Ye. 56-7-61/66
 TITLE: Observation of the Anomalous Decay of Charged Particles in the
 Wilson Chamber. (Nablyudenie anomal'nogo raspada zaryazhenncy
 chastitsy v kamere Vilsona, Russian)
 PERIODICAL: Zhurnal Eksperim. i Teoret. Fiziki, 1957, Vol 33, Nr 7, pp 303-303
 (U.S.S.R.)
 ABSTRACT: A slow particle with a more than 20-fold ionization enters the
 WILSON chamber (observation took place in the Elbrus Laboratory)
 and decays, on which occasion it emits a positive particle with
 a momentum of 352^{+94}_{-61} MeV/c at an angle of 95° . At present it is
 presumed that the decay of a particle which is heavier than a
 K-meson, was observed. (With 1 Illustration).
 ASSOCIATION: Physical Institute of the Georgian Academy of Sciences of the
 U.S.S.R. (Institut fiziki Akademii nauk Gruzinskoy S.S.R.)
 PRESENTED BY:
 SUBMITTED: 19.4.1957
 AVAILABLE: Library of Congress
 Card 1/1

ROMANTSEVA, L.M.; KOZLOVA, I.V.

Nikolai Nikolaevich Zinin; 75th anniversary of his death. Khim. v
shkole 10 no.5:17-26 S-O '55. (MIRA 8:11)
(Zinin, Nikolai Nikolaevich, 1812-1880)

ROMANTSEVA, L.M.; KOZLOVA, I.V.

Tagged atom method and its application. Khim.v shkole 10 no.3:
3-17 My-Je '56. (MLRA 9:8)
(Radioactive tracers)

BALUDA, V.P.; CHERNAYA, V.V.; MALIAROVSKIY, V.N.; TSYNKALOVSKIY, I.B.;
ROMANTSEVA, T.B.

Functional state of the blood coagulation system in healthy subjects.
Probl. gemat.i perel. krovi 6 no.1:59-61 '61. (MIRA 14:2)
(BLOOD—COAGULATION)

NOTKINA, M.A.; SOLODOVNIK, S.M.; BARANOVA, L.L.; LUSHINA, V.K.;
ROMANTSEVA, T.I.

Increasing the sensitivity of the determination of impurities in
pure metals. Zav.lab. 28 no.2:176-177 '62. (MIRA 15:3)

1. Gosudarstvennyy nauchno-issledovatel'skiy i proyektnyy
institut redkometallicheskoj promyshlennosti.
(Metals--Spectra)

HAZAROVA, Z.F.; BATOG, A.Ye.; YENAL'YEV, V.D.; ROMANTSEVICH, M.K.

Condensation of tertiary amyl hydroperoxide with some
carbonyl compounds. Zhur. ob. khim. 34 no.7:2430-2432
Jl '64 (MIRA 17:8)

1. Ukrainskiy nauchno-issledovatel'skiy institut plastmass,
Donetsk.

ROMANTSEVICH, M. K.

CZECH

✓ Condensation of olefin oxides with halogen compounds of antimony. M. S. Malinovsky and M. K. Romantsevich (L. Franko State Univ., Lvov). *Soviet Chemistry* (Engl. transl. from *Khim.* 2, 1366-9 (1963)). Satn. of 14.8 g. SbCl_3 with 28 g. ethylene oxide, with cooling, followed by 3 days at room temp. gave after evapn. of excess oxide 23 g. oil, which decomp. at 120° , identified as $\text{Sb}(\text{OCH}_2\text{CH}_2\text{Cl})_3$, $d_4 1.7604$; heated with H_2O it gave Sb_2O_3 and $\text{ClCH}_2\text{CH}_2\text{OH}$. Letting 14.27 g. SbCl_3 and 5.51 g. ethylene oxide stand in MePh 24 hrs, gave after evapn. 10.78 g. viscous liquid which solidified on standing, m. 67° , identified as crude $(\text{ClCH}_2\text{CH}_2\text{O})_3\text{SbCl}$. Treatment of 8.48 g. oxide with 18.04 g. SbCl_3 similarly gave crude $(\text{ClCH}_2\text{CH}_2\text{O})_3\text{SbCl}$, m. $60-70^\circ$. Propylene oxide (12 g.) and 8.07 g. SbCl_3 gave after 3 days 14.23 g. oily $\text{C}_3\text{H}_7\text{O}_2\text{Cl}_2\text{Sb}$, $d_4 1.4944$, which is also decomposed by H_2O , EtOH , AcOH . Epichlorohydrin and SbCl_3 gave after 2 days $\text{C}_3\text{H}_7\text{O}_2\text{Cl}_2\text{Sb}$, oil, $d_4 1.8451$. Treatment of 12 g. ethylene oxide with 18.08 g. SbBr_3 in MePh gave after standing overnight 24.69 g. oily $\text{C}_2\text{H}_5\text{O}_2\text{Br}_2\text{Sb}$, $d_4 2.3181$, which is readily attacked by H_2O , yielding $\text{HOCH}_2\text{CH}_2\text{Br}$. Reaction of ethylene oxide with BiCl_3 at 60° in MePh gave a sol. product, $(\text{ClCH}_2\text{CH}_2\text{O})_3\text{Bi}$, and an insol. product, $(\text{ClCH}_2\text{CH}_2\text{O})_2\text{BiCl}$. G. M. K.

ROMANTSEVICH, M. K.:

Romantsevich, M. K.: "The interaction of alpha-oxide compounds containing organochlorosilanes." Min Higher Education USSR. L'vov State U imeni Ivan Franko. Chair of Organic Chemistry. L'vov, 1956. (Dissertation of the Degree of Doctor of Chemical Science)

SC: Knizhnaya letopis', No 27, 1956. Moscow. Pages 94-109; 111.

ROMANTSEVICH, M. K.

✓ Synthesis and transformations of oxygen-containing organosilicon compounds. III. Reactions of propylene oxide with alkylarylsilanes. M. F. Shostakovskii.

M. S. Malinovskii, M. K. Romantsevich, and D. A. Kochkin (N. D. Zelinskii Inst. Org. Chem., Acad. Sci. U.S.S.R., Moscow). *Izv. Akad. Nauk S.S.S.R., Otdel. Khim. Nauk* 1956, 632-4; *cf. C.A.* 50, 11235a. To 29.04 g. propylene oxide (I) was added 75.36 g. iso-BuEtSiClH keeping the temp. at 30-5°; after 8 hrs. at 35-45° and standing overnight, the mixt. gave 84.9% iso-BuEtSiHOCH₂CHMeCl, b. 60-60.5°, *n*_D²⁰ 1.4330, *d*₄²⁰ 0.9181. Similarly were obtained (b.p., *n*_D²⁰, and *d*₄²⁰ given): 76.3% MeSiH(OCH₂CHMeCl), b. 90.5-1°, *n*_D²⁰ 1.4359, *d*₄²⁰ 0.9185; 77.1% Me₂Si(OCH₂CHMeCl)₂, b. 75.5-8°, *n*_D²⁰ 1.4378, *d*₄²⁰ 0.9188; 60.8% EtSiH(OCH₂CHMeCl), b. 85.2-8.4°, *n*_D²⁰ 1.4406, *d*₄²⁰ 0.9230; 72% CH₃CHSiHOCH₂CHMeCl, b. 140.5-1°, *n*_D²⁰ 1.4537, *d*₄²⁰ 1.1457. Shaking EtSiH(OCH₂CHMeCl) with concd. HNO₃ and heating 1 hr. at 50° gave MeCHClCO₂H. The results indicate that the ring of I opens at the primary C atom. Also in *Bull. Acad. Sci. U.S.S.R. Div. Chem. Sci.* 1956, 639-41 (Engl. translation). G. M. K.

Romantsevich, M. K.

Reaction of epichlorohydrin and methyl glycidyl ether with organochlorosilanes. M. S. Malinowski and M. K. Romantsevich (State Univ., Lvov). *Zhur. Obshchei Khim.* 27, 1080-5 (1957); *cf. C.A.* 49, 4605f. Chlorosilanes with epichlorohydrin give ring opening at the secondary C atom. Addn. of 29.76 g. epichlorohydrin to 14.95 g. MeSiCl_2 at 30-5° and heating 10 hrs. at 60° gave 77.3% $\text{MeSi}[\text{OCH}(\text{CH}_2\text{Cl})]_2$ (I), b. 202-4°, n_D^{20} 1.4012, d_4^{20} 1.3804. Similarly were prepd.: 62.2% $\text{Me}_2\text{SiOCH}(\text{CH}_2\text{Cl})_2$, b. 105-7°, 1.4394, 1.0762; 76.3% $\text{Me}_2\text{Si}[\text{OCH}(\text{CH}_2\text{Cl})]_2$, b. 135-7°, 1.4771, 1.2791; 87.5% $\text{Me}_2\text{SiH}[\text{OCH}(\text{CH}_2\text{Cl})]_2$, b. 132°, 1.4802, 1.3124; 73.5% $\text{Et}_2\text{Si}[\text{OCH}(\text{CH}_2\text{Cl})]_2$, b. 136-8°, 1.4786, 1.2352; 77.5% $\text{EtSi}[\text{OCH}(\text{CH}_2\text{Cl})]_2$, b. 191-3°, 1.4915, 1.3634; $\text{EtSiH}[\text{OCH}(\text{CH}_2\text{Cl})]_2$, b. 139°, 1.4790, 1.2805. I with $\text{Na}_2\text{Cr}_2\text{O}_7$ in aq. H_2SO_4 gave $\text{CO}(\text{CH}_2\text{Cl})_2$. Reaction with Et_3SiCl failed to take place. Me glycidyl ether (27.43 g.) with 14.95 g. MeSiCl_2 gave 29.3% $\text{Me}_2\text{Si}[\text{OCH}(\text{CH}_2\text{Cl})\text{CH}_2\text{OMe}]_2$, b. 170-70.5°, 1.4579, 1.2054. Similarly were prepd.: 71.4% $\text{Me}_2\text{Si}[\text{OCH}(\text{CH}_2\text{Cl})\text{CH}_2\text{OMe}]_2$, b. 119.5-20°, 1.4495, 1.1321; 77.1% $\text{Me}_2\text{SiOCH}(\text{CH}_2\text{Cl})\text{CH}_2\text{OMe}$, b. 86.5-8°, 1.4273, 0.9956; 76% $\text{MeSiH}[\text{OCH}(\text{CH}_2\text{Cl})\text{CH}_2\text{OMe}]_2$, b. 118.5-20°, 1.4495, 1.1491; 73.2% $\text{EtSi}[\text{OCH}(\text{CH}_2\text{Cl})\text{CH}_2\text{OMe}]_2$, b. 181-3°, 1.4599, 1.1968; 66.4% $\text{Et}_2\text{Si}[\text{OCH}(\text{CH}_2\text{Cl})\text{CH}_2\text{OMe}]_2$, b. 137-8°, 1.4588, 1.1202; 73.5% $\text{EtSiH}[\text{OCH}(\text{CH}_2\text{Cl})\text{CH}_2\text{OMe}]_2$, b. 128-30°, 1.4513, 1.1396. The structure of the products was confirmed by oxidation as above to $\text{MeOCH}_2\text{COCH}_2\text{Cl}$ and reduction of this with Zn-AcOH to MeOCH_2Ac , b. 117°, d_4^{20} 0.9558. The previous report by Andrianov, *et al.* (*C.A.* 50, 6302i), as to orientation of the reaction of epichlorohydrin with chlorosilanes is severely criticized. G. M. K.

4
2 May
3

Distr: 4E1j/4E3d/4E2c(j) KM

LOMANITSEVICH 7/1/46

Reaction of ethylene oxide and cyclohexene oxide with organochlorosilanes. M. G. Malinovsky and L. K. Romanova (State Univ., Lvov). *Zhur. Obshch. Khim.* 27, 1873-6 (1957).—Reaction of olefin oxides with chlorosilanes yields 2-chloroalkoxy derivs. Into 25.8 g. EtSiHCl_2 in dry Et_2O was introduced dry ethylene oxide yielding 80% $\text{EtSiH}(\text{OCH}_2\text{CH}_2\text{Cl})_2$, b. 68.5–8°, n_D^{20} 1.4464, d_4^{20} 1.1523. Similarly prepd. were: 75.7% $\text{Et}_2\text{SiH}(\text{OCH}_2\text{CH}_2\text{Cl})_2$, b. 50°, 1.4349, 0.9287; 79.7% $\text{ClCH}_2\text{SiMe}_2(\text{OCH}_2\text{CH}_2\text{Cl})_2$, b. 109°, 1.4671, 1.2671; 75.5% $\text{Cl}_2\text{CHSiMe}_2(\text{OCH}_2\text{CH}_2\text{Cl})_2$, b. 117–18°, 1.4798, 1.3454; 68.3% $\text{Et}_2\text{Si}(\text{OCH}_2\text{CH}_2\text{Cl})_2$, b. 123–4°, 1.4597, 1.2369; 75.3% $\text{Et}_2\text{Si}(\text{OCH}_2\text{CH}_2\text{Cl})_2$, b. 96–7°, 1.4528, 1.1011; 87.5% $\text{Et}_2\text{P}(\text{OCH}_2\text{CH}_2\text{Cl})_2$, b. 98–9°, 1.5043, 1.0522; 78.4% $\text{Me}_2\text{P}(\text{OCH}_2\text{CH}_2\text{Cl})_2$, b. 71–2°, 1.4331, 0.9472; 59.2% $\text{Me}_2\text{Si}(\text{OCH}_2\text{CH}_2\text{Cl})_2$, b. 72–3°, 1.4593, 1.3814. Similar reaction of cyclohexene oxide, requiring 10 hrs. at 60° for completion, gave: (*c*- $\text{ClC}_6\text{H}_{11}\text{O}$) $_2\text{SiMe}_2$, 70.3%, b. 211–14°, 1.4971, 1.1808; 73.4% (*c*- $\text{ClC}_6\text{H}_{11}\text{O}$) $_2\text{SiMe}_2$, b. 152–4°, 1.4814, 1.1178; 68.7% (*c*- $\text{ClC}_6\text{H}_{11}\text{O}$) $_2\text{SiHMe}$, b. 60–1.5°, 1.4565, 0.9931; 75.2% (*c*- $\text{ClC}_6\text{H}_{11}\text{O}$) $_2\text{SiHMe}$, b. 156–7°, 1.4865, 1.1404; 94.2% (*c*- $\text{ClC}_6\text{H}_{11}\text{O}$) $_2\text{Si}$, m. 47–51°. The chloroethyl derivs. are rapidly hydrolyzed by H_2O especially in the presence of acids. The reaction with ethylene oxide has an appreciable induction period.

G. M. Kosolapoff

5
4E4
4E3d
4E2c (J)
2may

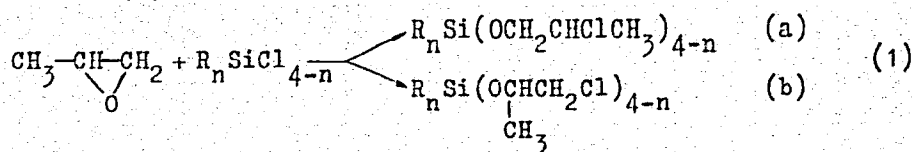
5 (3)

AUTHORS: Malinovskiy, M. S., Romantsevich, E. K. SOV/79-29-3-29/61

TITLE: On the Reaction of Propylene Oxide With Organochlorosilanes
(O vzaimodeystvii okisi propilena s organokhlorsilanami)

PERIODICAL: Zhurnal obshchey khimii, 1959, Vol 29, Nr 3, pp 888-891 (USSR)

ABSTRACT: In the present paper the synthesis of eight β -chloropropoxy-organosilanes hitherto not published is described and the way of cleavage of the oxide ring in the propylene oxide on the reaction with organochlorosilanes is investigated. This reaction (1) may proceed in two directions:



R. Sauer and W. Patnode (Ref 2) suggested the scheme (2) for the analogous reaction with ethylene oxide. As the chlorosilanes always contain traces of HCl, the ethylene oxide is transformed by them into the ethylene chlorohydrin. The latter reacts with organochlorosilanes according to scheme (3)

Card 1/3

On the Reaction of Propylene Oxide With Organo-
chlorosilanes

SOV/79-29-3-29/61

under regeneration of the hydrogen chloride which, for its part, reacts according to scheme 2 with the ethylene oxide. The propylene oxide reacts with the organochlorosilanes in the same way, only its oxide ring is cleaved with HCl in two directions, to the primary and secondary carbon. This yields a mixture of two chlorohydrins, (a) and (b) (Scheme 4), with considerable predominance of the isomer which corresponds to direction (a) (Ref 3, according to Petrov). For this reason the second step, the reaction of chlorohydrin with the organochlorosilanes, proceeds in two directions (5) and (6). On the oxidation of the hydrolysis products of the ethers obtained with nitric acid (Ref 1) a small amount of α -chloropropionic acid is formed which permits reaction (6) but indicates that the reaction of the propylene oxide with organochlorosilanes chiefly causes the formation of isopropylidene compounds corresponding to formula (b) in scheme (1). This assumption was supported by the oxidation of the hydrolysis products of the ethers with nitric acid and the chromium mixture. In the first case the α -chloropropionic acid resulted in low yield and in the second case

Card 2/3

On the Reaction of Propylene With Organo-
chlorosilanes

SOV/79-29-3-29/61

chloroacetone was obtained in good yield which confirms equation (5) or the direction (b) in scheme (1). Thus in the above reaction the isomer with the oxygen at the secondary carbon atom prevails in the mixture of isomers obtained. There are 1 table and 5 references, 3 of which are Soviet.

ASSOCIATION: Dnepropetrovskiy gosudarstvennyy universitet (Dnepropetrovsk State University)

SUBMITTED: February 1, 1958

Card 3/3

1. *Phragmites* and *Spartina* spp. are common in the marshes. 10/10/77
 2. *Phragmites* and *Spartina* spp. are common in the marshes. 10/10/77

1. Izvorna izjava o ispitivanju i ispitivanju institucije i institucije
izvorne, izvorne.

ROMANTSEVICH, M.K. [Romantsevyeh, M.K.]; SHOLOGON, I.M. [Sholohon, I.M.];
BARANOVSKAYA, N.F. [Baranovs'ka, N.F.]; SIRENKO, N.N.

Synthesis of dicyclopentadienedicarboxylic acid. Khim. prcm. [Ukr.]
no.1:20-22 Ja-Mr '65. (MIRA 18:4)

SOMOKINA, A.N.; BATOG, A.Ye.; ROMANTSEVICH, M.X.

Parent-containing peresters. Zhur. org. khim. 1 no.11:2050.
2051 N '65. (M.A. 18.12)

1. Submitted October 24, 1964.

L 21778-66 EWT(m)/EWP(i)/EWP(t) IJP(c) JD

ACC NR: AP6002511

SOURCE CODE: UR/0286/65/000/023/0018/0018

AUTHORS: Shologon, I. M.; Kapitonov, V. M.; Romantsevich, M. K.

ORG: none

TITLE: A method for obtaining bicyclopentadienyl titanium derivatives, containing silicon. Class 12, No. 176583²¹ /announced by Ukrainian Scientific Research Institute for Plastics (Ukrainskiy nauchno-issledovatel'skiy institut plastmass) /

SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 23, 1965, 18

TOPIC TAGS: organotitanium compound, organosilicon compound, organic chemistry, titanium, silicon, sodium

ABSTRACT: This Author Certificate presents a preparative method for obtaining bicyclopentadienyl titanium derivatives containing silicon. The sodium derivatives of cyclopentadienyltrialkyl (aryl) silanes are treated with titanium tetrachloride at -30 to -40C in an organic solvent, e.g., tetrahydrofuran.

SUB CODE: 07/ SUBM DATE: 30Nov64

Card 1/1 ULR

UDC: 547.419.5/6:514.721.07

BATOG, A.Ye.; LATARSKAYA, I.M.; BOCHAROVA, Yu.Ye.; YENAL'YEV, V.D.;
ROMANISEVICH, M.K.

Synthesis of peroxide and hydroperoxide of tertiary butyl.
Ukr.khim.zhurn. 31 no.2:207-208 '65. (MIRA 18:4)

1. Ukrainskiy nauchno-issledovatel'skiy institut plasticheskikh
mass, Donetsk.

L 26370-66 EWP(j)/EWT(m)/T RM/JD

ACC NR: AF6011197

SOURCE CODE: UR/0413/66/000/006/0028/0028

INVENTOR: Batog, A. Ye.; Mashnenko, O. M.; Romantsevich, M. K.

ORG: none

TITLE: A method for producing peroxide compounds containing silicon. Class 12, No. 179770 [announced by Ukrainian Scientific Research Institute of Plastics (Ukrain-skiy nauchno-issledovatel'skiy institut plastmass)]

SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 6, 1966, 28

TOPIC TAGS: silicon plastic, peroxide, halogenated organic compound, silane

ABSTRACT: This Author's Certificate introduces a method for producing peroxide compounds containing silicon in which the peroxide group is not directly linked to the silicon atom. Alkyl(aryl)peroxyalkanols are interacted with organic silane halides in the presence of a base in an inert solvent, e. g. hexane, at a temperature of -20 to +40°C.

SUB CODE: 07,11/

SUBM DATE: 30Nov64/

ORIG REF: 000/

OTH REF: 000

UDC: 547.419.5-39.07

Card 1/1

L 44571-66 EWT(m)/EWP(j) IJP(c) RM

ACC NR: AP6015676

SOURCE CODE: UR/0413/66/000/009/0077/0077

INVENTOR: Shologon, I. M.; Moshinskiy, L. Ya.; Romantsevich, M. K. 28
B

ORG: none

TITLE: Method of obtaining organosilicon resins. ¹⁵ Class 39, No.181297 ¹⁵
[announced by Ukrainian Scientific Research Institute of plastics
(Ukrainskiy nauchno-issledovatel'skiy institut plastmass)]

SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 9,
1966, 77

TOPIC TAGS: resin, organosilicon resin, organosilicon compound

ABSTRACT: An Author Certificate has been issued for a method of obtain-
ing organosilicon resins by the condensation of silicon organic com-
pounds with polydydric alcohols upon heating. To expand the variety of
initial compounds, alkoxysilylendomethylenetetrahydrophthalic anhydride
is suggested as the organosilicon compound. [Translation] [NT]

SUB CODE: 11/ SUBM DATE: 09Jul64/

Card 1/1 *28m*

UDC: 678.6:661.72

ACC-NR: AP6033182

SOURCE CODE: UR/0079/66/036/010/1846/1848

AUTHOR: Shologon, I. M.; Romantsevich, M. K.

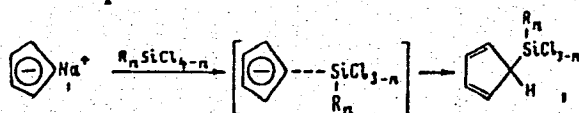
ORG: none

TITLE: Synthesis and reactions of silicon-containing derivatives of cyclopentadiene.
Part 1: Cyclopentadienylchlorosilanes

SOURCE: Zhurnal obshchey khimii, v. 36, no. 10, 1966, 1846-1848

TOPIC TAGS: organosilicon compound, silane, organosodium compound, *chemical synthesis, chemical reaction*

ABSTRACT: In an attempt to develop an effective method for preparing cyclopentadienylchlorosilanes, the condensation of cyclopentadienylsodium with silicon tetrachloride and organochlorosilanes was investigated. It is postulated that the condensation with chlorosilanes takes place via an ionic mechanism:



where R is an organic radical or chlorine, and n = 0, 1, 2, 3. Relatively high yields (60-70%) of cyclopentadienylchlorosilanes were obtained by adding a cyclopentadienylsodium suspension in toluene to an equimolar quantity of the initial chloro-

Card 1/2

UDC: 546.287

ACC NR: AP6033182

silane if the reaction temperature is maintained at -15 to 0°C. Some properties of the compounds obtained are shown in Table 1. Orig. art. has: 1 table.

Compound	Yield (%)	B.P. (p in mm)	n _D ²⁰	d ₄ ²⁰	MR _D	
					measured	calculated
C ₅ H ₅ SiCl ₃	58.3	46—48° (4)	1.5100	1.3632	43.79	44.77
C ₅ H ₅ Si(CH ₃)Cl ₂	72.5	47—49 (8)	1.4920	1.1622	44.73	44.83
C ₅ H ₅ Si(C ₂ H ₅)Cl ₂	63.9	76—78 (10)	1.4982	1.1530	49.12	49.34
C ₅ H ₅ Si(CH=CH ₂)Cl ₂	37.4	63—65 (8)	1.5070	1.1710	48.56	48.99
C ₅ H ₅ Si(C ₆ H ₅)Cl ₂	70.2	135—136 (6)	1.5666	1.2266	64.28	64.65
C ₅ H ₅ Si(CH ₃) ₂ Cl	60.2	54—56 (14)	1.4870	1.0048	45.43	45.75
C ₅ H ₅ Si(C ₂ H ₅) ₂ Cl	62.5	78—80 (17)	1.4675	0.9986	53.83	54.31
C ₅ H ₅ Si(CH=CH ₂)Cl	60.4	74—75 (23)	1.4970	1.0139	49.26	49.45
CH ₃						

SUB CODE: 07/ SUEM DATE: 11Sep65/ ORIG REF: 002/ OTH REF: 005

Card 2/2

ROMANTSOV, D.S.; IRLAKHMAN, Ya.A.

Regulations for the installation of electrical equipment in coal processing plants. Prom. energ. 18 no.3:40-42 Mr '63.

1. Trest "Kuzbassugleobogashcheniye" (for Romantsov).
2. Novosibirskoye otdeleniye Gosudarstvennogo proyektnogo instituta po proyektirovaniyu predpriyatiy elektropromyshlennosti (for Irlakhman).

(Coal preparation—Electric equipment)

SIDOROV, M.D.; BANDARENKO, Yu.A., inzh., retsenzent; MART'YANOV, E.V.,
inzh., retsenzent; ROMANTSOV, E.I., inzh., retsenzent; CHERNOUSOV,
N.P., inzh., retsenzent; GOFLIN, A.P., kand. tekhn. nauk, red.;
VASIL'YEVA, V.P., red.izd-va; SHCHETININA, L.V., tekhn. red.

[Handbook on air and gas blowing machines] Spravochnik po voz-
dukhoduvnym i gazoduvnym mashinam. Moskva, Mashgiz, 1962. 257 p.
(MIRA 15:12)

(Fans, Mechanical) (Air compressors) (Belts and belting)

ROMANTSOV, V.V., inzh.

Design and operation of thermal protection systems of once-
through type boilers. Elek. sta. 34 no.10:2-5.0 '63.
(MIRA 16:12)

ROMANTSOVA, G.I.

Effect of molecular interaction on the ultraviolet absorption spectra of aromatic compounds. Probl.fiz.khim. no.2:107-117 '59. (MIRA 13:7)

1. Laboratoriya stroeniya molekul Nauchno-issledovatel'skogo fiziko-khimicheskogo instituta Imeni L.Ya.Karpova.
(Aromatic compounds--Spectra)
(Spectrum, Ultraviolet)

ROMANTSOVA, G.I.

Spectrophotometric determination of small additions of aromatic
carboxylic acids in terephthalic acid. Zhur. prikl. spekt. 2
no. 3: 269-272 Mr '65. (MIRA 18:6)

Committee Secretariat

PHASE I BOOK EXPLOITATION SOV/4386
Moscow. Fiziko-khimicheskiy Institut

Problemy fizicheskoy khimii: Izv. 2 (Problema in Physical Chemistry: Transactions of the Institute, no. 2). Moscow, Gostekhizdat, 1959. 202 p. 1,000 copies printed.

Editorial Board: Ya. N. Varshavsky, Doctor of Chemical Sciences; O. S. Zhdanov, Doctor of Chemical Sciences; V. A. Kargin, Doctor of Chemical Sciences; S. S. Nedvedev, Doctor of Chemical Sciences (Resp. Ed.); S. S. Nedvedev, Academician; S. Ya. Papentsev, Doctor of Chemical Sciences; V. M. Chervulinenko, Candidate of Chemical Sciences; V. S. Chaslava (Editorial Secretary). Ed.: Ye. O. Shipak.

PURPOSE: This collection of articles is intended for physical chemists.

CONTENTS: The collection is the second issue of the Transactions of the Scientific Research Institute of Physical Chemistry named L. Ya. Karpov. It contains 17 articles which provide Card 1/5

Tsarkin, M. I., N. M. Morozov, V. M. Puzov (Deceased), L. O. Apel'duga, L. I. Lukyanova, and V. A. Deniskin. The Oxidation of Ammonia Over a Nonplatinum Catalyst. 18

Dubinskii, S. Ya., S. A. Kamenetskaya, Ye. I. Orlov, A. V. Pankratov, N. M. Kozlov, I. N. Pospelov, A. Ya. Apin, I. M. Shklyarskiy, N. A. Slonimskiy, and V. M. Chervulinenko. Kinetics of Decomposition, and the Explosion of Oxone 27

Borichuk, Yuro (Japan). How to Find the Kinetic Equation of a Reversible Reaction 39

Kolozhkin, Ya. M. The Effect of the Specific Adsorption of Hydrogen on the Kinetics of Hydrogen Evolution and the Structure of the Metal-Solution Boundary 50

Vartanyan, Ya. M. The Nature and Mechanism of Electro-Palladium Nitrogen Exchange 61

Zhdanov, O. S. Crystallochemical Data on the Nature of the Mutual Effect of Atoms 97

Mezharinov, A. I. Investigation of the Effect of Inter-molecular Interaction on the Ultraviolet Absorption Spectra of Aromatic Compounds 107

Shagina, Ye. I., V. S. Kuznetsov, and B. P. Omonov. Investigation of Equilibrium in the System of Hydrogen at High Temperatures and the Dependence of the Free Energy of H₂ Formation on Its Composition and Structure 118

Borichuk, A. D., M. A. Pankratov, L. A. Dzhurav, L. I. Shklyarskiy, and V. M. Chervulinenko. Study of the Field of Forces of Distance from a Catalytic Indicator with COO as a Powerful Source of γ Radiation 132

Kozlov, Y. K., B. O. Vasil'yev, and N. N. Tuntsevskiy. Study of the Ionization and Dissociation of n-Octane and n-Nonane Molecules by the Method of Bombardment with Quasi-Monochromatic Electrons 145

Petrovkin, A. S. Radiation-Chemical Effects in Solid Inorganic Salts 163

Melamed, M. P., A. V. Zil'ber, and R. V. Demidovskiy. Radiation-Chemical Chlorination of Benzene 169

Engelhardt, M. A., Ye. V. Barabko, and L. I. Fortishova. Course of the Process of Benzene Oxidation in an Aqueous Solution under the Action of Radiation 177

Kozlov, S. (Gerasimovskiy), Ye. V. Barabko, L. I. Fortishova, and M. A. Engeltshova. Study of the Composition of Benzene Radicals Formed During the Radiolysis of Benzene in an Aqueous Solution 193

Smirnov, V. A., and G. A. Gol'df. The Problem of the Phase Composition of the System H₂O-NH₃-NaOH at Low Temperatures 189

Zhdanov, V. D., and A. A. Zhdanovskaya. Sensitization of the Radiolytic Oxidation of Lanthanum Dyes 191

5.3700

77390

SOV/79-30-1-51/78

AUTHORS: Romantsevich, M. K., Malinovskiy, M. S.

TITLE: Concerning the Reaction of 2,3-Epoxy-1-propanol With
Organochlorosilanes

PERIODICAL: Zhurnal obshchey khimii, 1960, Vol 30, Nr 1, pp 232-
234 (USSR)

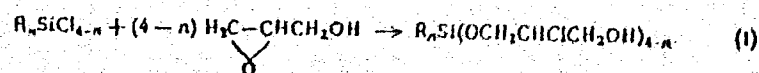
ABSTRACT: The authors reported previously (Izv. AN SSSR, OKhN.,
1956, p 232; ZhOKh, 1957, Vol 27, pp 1680 & 1873) the
reaction of various asymmetric epoxides (ethylene oxide,
propene oxide, 1-chloro-2,3-epoxypropane, methyl ether
of 2,3-epoxy-1-propanol, etc.) with organochlorosilanes.
Particular attention was paid to the way in which the
epoxy ring opened in the above reactions. The present
study deals with the reaction of 2,3-epoxy-1-propanol
(glycidol) with organochlorosilanes which, according
to Andrianov, et al. (Izv. AN SSSR, 1955, OKhN., 1955,
p 531) proceeds as follows:

Card 1/4

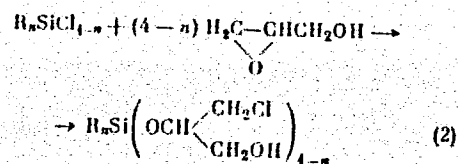
Concerning the Reaction of 2,3-Epoxy-1-propanol With Organochlorosilanes

77390

SOV/79-30-1-51/78



The authors, however, expected this reaction to proceed similarly to that of epichlorohydrin or methyl ether of glycidol, and the epoxy ring to be cleaved at the carbon atom with the lesser number of hydrogen atoms:



This assumption was confirmed by the reactions of glycidols with CH_3SiCl_2 , $(CH_3)_2SiCl_2$, $(CH_3)_3SiCl$, and CH_3SiHCl_2 . For example, glycidol and $(CH_3)_2SiCl_2$ were added to each other by means of two dropping funnels so

Card 2/4

Concerning the Reaction of 2,3-Epoxy-1-propanol With Organochlorosilanes

77390

SOV/79-30-1-51/78

as to maintain at all times a slight excess of glycidol and to keep the temperature below 30°C . The mixture was then heated for 4 hr at $50-60^{\circ}\text{C}$, and the reaction products were distilled under vacuum. The reaction gave di-(α -chloro- γ -hydroxyisopropoxy)-dimethylsilane (yield 59.4%; bp $96-97^{\circ}\text{C}$ at 4 mm; n_D^{20} 1.4734).

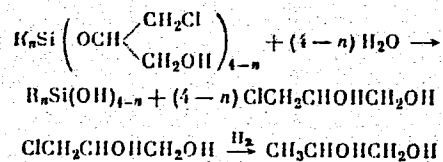
Similarly, α -chloro- γ -hydroxyisopropoxytrimethylsilane (yield 82.7%; bp $82.5-83^{\circ}$ at 13 mm; n_D^{20} 1.4342) and tri-(α -chloro- γ -hydroxyisopropoxy)methylsilane (yield 36.5%; bp $132-133^{\circ}\text{C}$; n_D^{20} 1.4840) were obtained from $(\text{CH}_3)_3\text{SiCl}$ and CH_3SiCl_3 , respectively. The structure of the above was confirmed by hydrolysis and subsequent reduction; in all three instances, α -propylene glycol was obtained in high yield, and it could have been formed only from the products of reaction (2) according to the reaction:

Card 3/4

Concerning the Reaction of 2,3-Epoxy-1-propanol With Organochlorosilanes

77390

SOV/79-30-1-51/78



There are 1 table; and 4 references, 1 German, 3 Soviet.

ASSOCIATION: Lvov Zoological and Veterinary Institute (L'vovskiy zooveterinarnyy institut)

SUBMITTED: November 15, 1958

Card 4/.

INDEX AND CROSS INDEX																									
PROCESSING AND PREPARATION INDEX													INDEX AND CROSS INDEX												
<i>Influence of the Addition of Titanium to Aluminium-Magnesium Alloys.</i> (G. Komovski, I. Romantsov, and A. Maksimov (<i>Lepkie Metally</i> (Light Metals), 1936, (12), 40-43).—[In Russian.] An investigation of the tensile strength of cast aluminium alloys with magnesium (2-12%), manganese (0.1-0.8%), and titanium (0.1-0.5%) showed that an alloy with magnesium 8, titanium 0.4, and manganese 0.4% had the maximum tensile strength. — D. N. S.																									
ASAC-35.4 METALLURGICAL LITERATURE CLASSIFICATION																									
STANDARD													STANDARD												

ALPHABETIC INDEX																									
A	B	C	D	E	F	G	H	I	J	K	L	M	N	O	P	Q	R	S	T	U	V	W	X	Y	Z
PROCESSES AND PROPERTIES INDEX 2 *Influence of the Addition of Titanium to Aluminum Magnesium Alloys. G. Komowski, I. Romantsov, and A. Maksimov (<i>Legkie Metally (Light Metals)</i>, 1936, (12): 40-43). [In Russian.] An investigation of the tensile strength of cast aluminum alloys with magnesium (2-12%), manganese (0.1-0.6%), and titanium (0.1-0.5%) showed that an alloy with magnesium 8, titanium 0.4, and manganese 0.4% had the maximum tensile strength. D. N. S.																									
ALPHABETIC INDEX																									
A	B	C	D	E	F	G	H	I	J	K	L	M	N	O	P	Q	R	S	T	U	V	W	X	Y	Z
DETAILED LITERATURE CLASSIFICATION																									

ROMANTSOV, I. I.

PA 21T114

USSR/Physics
Polarization
Light - Polarization

Sep 1946

"Photographing the Tropopause in Polarized Light," I. I. Romantsov, I. A. Khvostikov, Laboratory of Atmospheric Optics, Institute of Theoretical Geophysics, Academy of Sciences of the USSR, 3 pp

"Comptes Rendus (Doklady)" Vol LIII, No 8

A discussion is made of several investigations into the tropopause (the area between stratosphere and troposphere) which were performed by means of a powerful searchlight beam directed up 600 to the horizon and by a camera placed 7.6 km away. Four graphs are given

21T114

USSR/Physics (Contd)

Sep 1946

Polarization
Light - Polarization

showing the dependence of brightness (degree of polarization) upon height in km above ground.

21T114

ROMANTSOV, I. I.

"Searchlight Sounding of the Atmosphere in Polarized Light." Sub 12 Mar 47,
Inst of Theoretical Geophysics, Acad Sci USSR

Dissertations presented for degrees in science and engineering in Moscow
in 1947

SO: Sum No. 457, 18 Apr 55

Can. Physics - Smith, S.

ROMANTSOV, I. I.

USSR/Physics

Atmosphere

Sep/Oct 48

"Study of the Troposphere Using a Searchlight Beam,"
I. I. Romantsov, Lab of Atm Opt, Inst Theoretical
Geophys, Acad Sci USSR, 17 pp

"Iz Ak Nauk SSSR, Ser Geog i Geofiz" Vol XII, No 5

PA 53/49T93
Gives results of atmospheric observations at altitudes up to 13 km. Constructed curves showing degree of polarization as a function of the height of the dispersed layer of the atmosphere for 13 nights. Experimental data obtained determines nature of light dispersion at various altitudes. Specifically, made definite assumptions on the nature of dispersed particles suspended in an atmosphere.
53/49T93

USSR/Physics (Contd)

Sep/Oct 48

atmospheric layer for 4 - 9 km altitudes. Observed tropopause and negative polarization. Submitted by Acad I. S. Leybenzon, 10 Oct 46.

53/49T93

ROMANTSOV, V.V.
DEYANOV, V.A., inzhener; ROMANTSOV, V.V., inzhener

Improving the supply circuit for automatic controllers of an
electromechanical system. Elek.sta.26 no.8:53-54 Ag'55.
(MLRA 8:12)

(Automatic control) (Electric circuits)